\int	BHARATH COACHING CENTRE	
7 th CB	SE Physical and Chemical Change	Total: 30
Sciend	ce	Time: 1.30 hrs
SECTIO	N – A	3 × 1 = 7
	ill in the blanks	
1.	Burning of magnesium ribbon forms	
2.	The process of depositing layer of zinc on iron is calle	d
II. State whether the given changes are either physical or chemical change		
3.	Photosynthesis	
4.	Burning of coal	
5.	Melting of candle	
6.	Making lemon juice	
7.	Making jewelry from gold	
<u>SECTIO</u>	<u>N – B</u>	3 × 2 = 6
8.	Suggest two methods to prevent rusting.	
9.	Define Crystallization.	
10.	What is browning of vegetable?	
<u>SECTION – C</u>		4 X 3 = 12
11. Define rusting. Write the chemical equation.		
12. Differentiate between physical and chemical change. Give an example for each.		

- 13. Why does iron rust but stainless steel doesn't?
- 14. Differentiate between element and compound.

<u>SECTION – D</u>

15. What will happen if we drop iron nails into copper sulphate solution? Write the reaction between copper sulphate and iron nails.

 $1 \times 5 = 5$