## BHARATH COACHING CENTRE

8<sup>th</sup>CBSE Science Total: 40

Chemical Effects of Electric Current Time: 1.30 hrs

 $\underline{\mathsf{SECTION} - \mathsf{A}} \qquad \qquad \mathsf{5} \,\mathsf{X} \,\mathsf{1} = \mathsf{5}$ 

1. Which of the following is a electrolyte used for electroplating silver? b) CuSO<sub>4</sub> c) AgNO<sub>3</sub> d) HNO<sub>3</sub> a) AgSO<sub>4</sub> 2. Distilled water is a \_\_\_\_\_ a) Conductor b) Insulator c) Semiconductor d) None of these 3. Anions are attracted to the \_\_\_\_\_. a) Cathode b) Anode c) Electrolyte d) container 4. The electrode connected to positive terminal is called a) Anode b) cathode c) Strap d) Rod 5. Flow of charged particles constitute a) Electricity b) Electron c) Proton d) Electrolysis

 $\underline{\mathsf{SECTION} - \mathsf{B}} \qquad \qquad \mathsf{5} \,\mathsf{X} \,\mathsf{2} = \mathsf{10}$ 

- 6. What is electroplating? Write its uses.
- 7. Define electrical conductivity.
- 8. Explain electro refining.
- 9. Name two metals widely used in automobile industry for protection from corrosion of metal parts.
- 10. List out the electroplated objects in our daily life.

 $\underline{\mathsf{SECTION} - \mathsf{C}} \qquad \qquad 5 \,\mathsf{X} \,\mathsf{3} = \mathsf{15}$ 

- 11. Why do impurities dissolved in water increase its conductivity?
- 12. What is anode mud?
- 13. Describe the uses of electrolysis.
- 14. Describe electroplating with examples.
- 15. Draw the general setup for electroplating and explain it.

2 X 5 = 10SECTION – D

- 16. Give reasons for the following.
  - a) Operating electrical appliances with wet hands is very dangerous.
  - b) Pure water is a poor conductor of electricity.
  - c) Cathode and anode has to be placed in distance from one another.
- 17. With respect to electro refining of copper, answer the following questions.
  - a) Which object is used as the anode?
  - b) Name the electrolyte used.
  - c) What is used as cathode?

