## BHARATH COACHING CENTRE

8<sup>th</sup> CBSE Science Total: 40

Materials: Metals and Non-metals Time: 2.00 hrs

 $\underline{\mathsf{SECTION} - \mathsf{A}} \qquad \qquad \mathsf{8} \times \mathsf{1} = \mathsf{8}$ 

1. Name a soft metal.

- 2. Name a hard non-metal.
- 3. Name the metal which is used to galvanize steel.
- 4. Name a metal which is liquid in room temperature.
- 5. Name the non-metal which is used in pencil lead.
- 6. Name a metal which is stored under water to avoid combustion.
- 7. Name a non-metal which is liquid in room temperature.
- 8. Name a non-metal which conducts electricity.

 $\underline{\mathsf{SECTION} - \mathsf{B}} \qquad \qquad 3 \times 2 = 6$ 

- 9. Why does iron rust while steel doesn't?
- 10. Why should Na be stored under kerosene? What happens when it comes in contact with air?
- 11. Define rusting.

 $\underline{\mathsf{SECTION} - \mathsf{C}} \qquad \qquad 2 \times 3 = 6$ 

- 12. What is an alloy? Give two examples.
- 13. The metal 'X' easily combusts and is used in fireworks. When the metal ribbon is burnt it produces white powder 'Y'. 'Y' is a metal oxide and acidic in nature. Identify 'X' and 'Y'.

 $\frac{\text{SECTION} - C}{4 \times 5} = 20$ 

- 14. Define the given terms and give an example for each.
  - (a) Malleability (b) ductility (c) sonority (b) thermal conductivity (e) electrical conductivity
- 15. Write 5 uses each for metals and non-metals
- 16. Write the difference in physical properties of metals and non-metals.

## 17. Complete the reactions

- (a)  $S + O_2 \longrightarrow$
- (b) Cu + FeSO<sub>4</sub> →
- (c) C + O<sub>2</sub>
- (d) Cu + conc.  $H_2SO_4$
- (e) Zn + 2HCl →